Microwave-Assisted Formation of Sulfonium Photoacid Generators

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Efficient method for preparing sulfonium photoacid generators (PAGs) by microwave-assisted reaction of diarylsulfides in the presence of alkylaryliodonium salts. Microwave-assisted synthesis of the PAGs is significantly faster, reducing reaction time with less energy consumption. Reaction times using the microwave-assisted synthesis are in a range of from 90 to 420 faster than conventional thermal conditions. The photoacid quantum yields of several salts prepared by the microwave reaction were measured; the photoacid quantum yields of new sulfonium PAGs were in a range from 0.01 to 0.4 times greater than yields from conventional synthesis. Incorporating a nitro group in the structures of the sulfonium salts increased photoacid quantum yield, induced intersystem crossing and increased the efficiency of photoacid generation compounds.
OTHER PUBLICATIONS
* cited by examiner
1. NaBH₄-10% Silica, (mw), r.t.
2. HBr/AcOH (33%), reflux

1. NBS, PC, 60°C
2. n-BuLi (1.6M), -78°C to r.t

1. P(EO)₃, reflux
2. NaH, DMF r.t.

Δ = conventional heating
MW = microwave heating

Figure 1
\((\text{Ph)}_3\text{P}^+\text{CH}_3\text{Cl}^-\) 

\[
\text{CH}_2\text{Cl}_2, \text{rt} 
\]

\[
\begin{align*}
\text{8} & \quad 42\% \\
\text{9} & \quad \text{Pd(OAc)}_2, \text{N(Et)}_3, \\
& \quad \text{DMF,} 150 \text{ °C (MW)}
\end{align*}
\]

\[
\begin{align*}
\text{6} & \quad \text{Cu(II)Benzoate} \\
& \quad \text{Chlorobenzene} \\
& \quad 125 \text{ °C (Δ or MW)}
\end{align*}
\]

\[
\begin{align*}
\text{10} & \quad 99\% \\
\text{11} & \quad 97\% (\text{MW}) \\
& \quad 87\% (Δ)
\end{align*}
\]

Δ = conventional heating  
MW = microwave heating

\text{Figure 2}
Figure 3

\[ \text{C}_6\text{H}_{12}\text{I}\text{H}_21\text{C}_6\text{H}_{12}\text{NO}_2 \xrightarrow{\text{SH}, \text{Os}_2\text{CO}_3} \text{C}_6\text{H}_{12}\text{C}_6\text{H}_{12}\text{NO}_2 \]

\[ \text{PF}_6^- \]

\[ \text{Cu(II)Benzoate} \]

\[ \text{Chlorobenzene} \]

\[ 125^\circ\text{C} (\Delta \text{ or MW}) \]

\[ \text{73\%} \]

\[ \text{C}_6\text{H}_{12}\text{C}_6\text{H}_{12}\text{NO}_2 \]

\[ \text{PF}_6^- \text{S}^+ \]

\[ \text{51\%(MW)} \]

\[ \Delta = \text{conventional heating} \]

\[ \text{MW} = \text{microwave heating} \]
Figure 4

\[ \text{Br-S} \xrightarrow{1, \text{KOH, MeOH, rt}} \text{Br-S} \]

\[ \text{Cu(II)Benzoate Chlorobenzene 125 °C (\(\Delta\) or MW)} \]

\[ \text{Br-S} \xrightarrow{\text{PF}_6^-} \]

\[ \text{83\% (MW) 71\% (\(\Delta\))} \]

\(\Delta = \text{conventional heating}\)
\(\text{MW} = \text{microwave heating}\)
\[
R \begin{array}{c}
\text{S} \\
\text{PF}_6^-
\end{array}
\xrightarrow{	ext{Cu(II)Benzoate}}
R \begin{array}{c}
\text{S} \\
\text{PF}_6^-
\end{array}
\]

Chlorobenzene

125 °C (Δ or MW)

\[
\begin{array}{ccc}
17 & 19 & 21 \\
\text{R} = \text{CH}_3, & \text{CH}_3, & \text{CH}_3, \\
\text{H} & \text{H} & \text{H}
\end{array}
\]

18 72%(MW) 120s
20 98%(MW) 120s
22 88%(MW) 90s

Δ = conventional heating
MW = microwave heating

Figure 5
MW = microwave heating

Figure 6
MICROWAVE-ASSISTED FORMATION OF SULFONIUM PHOTOACID GENERATORS

This application claims the benefit of priority from U.S. Provisional Patent Application Ser. No. 61/044,722 filed Apr. 14, 2008 and is incorporated herein by reference.

The research was supported in part by the National Science Foundation under Grants # ECS-0621715 and ECS-0524533.

FIELD OF THE INVENTION

This invention relates to photoacid generators and, in particular, to methods, systems, apparatus and devices for preparing sulfonium salt photoacid generators (PAGs) using microwave-assisted reactions.

BACKGROUND AND PRIOR ART

Since their discovery, the use of photoacid-generators, PAGs, has been widely adopted by the polymer industry in coatings, paints, anticorrosives, and electronics as described in Crivello, J. C. Journal of Polymer Science Part A: Polymer Chemistry (1999), 37, pp. 4241-4254.

The use PAGs in epoxide photoresists has been extensively studied for two-dimensional (2D) and three-dimensional (3D) lithographic patterning induced by one-photon absorption (1PA) and two-photon absorption (2PA) as described in Jeon, S.; Malyarchuk, V.; Rogers, J. A.; Wiederrecht, G. P. Optics Express (2006), 14, pp. 2300-2308 and Kim, E. K.; Ekerdt, J. G.; Willson, C. G. Journal of Vacuum Science & Technology B (2005), 23, pp. 1515-1520.

Two-photon absorption (2PA) has been reported for a number of materials, exploiting the fact that the 2PA probability is directly proportional to the square of the incident light intensity (dw/dt) according to M. Goeppert-Mayer in 1931; while one-photon absorption bears a linear relation to the incident light intensity, dw/dt. This intrinsic property of 2PA leads to 3D spatial localization, important in fields such as optical data storage, fluorescence microscopy, and 3D micro-fabrication.

There have been demonstrations of successful two-photon microfabrication using commercial PAGs, as reported by Belfield, K. et al. in J. Phys. Org. Chem. 2000, 13, 837-849 and Belfield, K. et al. in J. Am. Chem. Soc., 2000, 122, 1217-1218 even though the 2PA cross section of these initiators is low according to Schaefer, K. et al. in J. Photochem. Photobiol., A, 2004, 162, 497-502. Two-photon 3D microfabrication with a novel PAG was reported to fabricate MEMs structures by Yu, T. et al. in Advanced Materials, 2003, 15, 517-521. To further advance a number of emerging technologies, there is a great need for photonic generators (PAGs) with higher 2PA cross sections.

The synthesis of a class of photoacid generators such as, triarylsulfonium salts, by photolysis of the diphenyliodonium counterpart in the presence of a diphenylsulfide was originally reported by Crivello et al. in Abstr. Pap. Am. Chem. Soc., 1978, 176, 8-8 and J. Org. Chem. 1978, 43, 3055-3058.

Microwave-facilitated synthesis has been the subject of substantial interest over the last decade, as reported by Lidstrom, P. et al. in Tetrahedron, 2001, 57, 9225-9283 and Nillsson, P. et al. in Microwave Methods in Organic Synthesis 2006, vol. 266, 103-104. Scipoli et al. in J. Org. Chem. 2008, 73, 4291-4294 recently reported microwave-assisted reaction times are 80 times faster than conventional heating reaction times, and are potentially more energy-efficient than conventional heating. There are no known reports of the microwave-assisted synthesis of sulfonium salts and no knowledge that microwave assisted synthesis would provide the advantage of reducing reaction times, which is extremely useful and cost efficient.


The present invention discloses a facile, microwave-assisted synthesis of triarylsulfonium salt photoacid generators (PAGs) suitable for use as two-photon absorbing (2PA) photoinitiators, in negative resists for photolithography or 3-D microfabrication, or in optical data storage.

SUMMARY OF THE INVENTION

A primary objective of the invention is to provide methods, apparatus and systems for an efficient method for preparing sulfonium photoacid generators (PAGs) by microwave-assisted reaction of diphenylsulfides in the presence of diphenyliodonium salts.

A second objective of the invention is to provide methods, apparatus and systems for microwave-assisted synthesis of the PAGs that is significantly faster, in a range from 90 to 420 times faster than conventional thermal conditions, reducing reaction time with energy consumption.

A third objective of the invention is to provide methods, apparatus and systems for microwave-assisted synthesis of the PAGs with photoacid quantum yields of sulfonium PAGs ranging from 0.01 to 0.4.

A fourth objective of the invention is to provide novel photoacid generators (PAGs) for use as 2PA photoinitiators in negative resists for three-dimensional (3D) microfabrication.

A fifth objective of the invention is to provide novel photoacid generators (PAGs) for use as 2PA photoinitiators in optical data storage.

A preferred method for synthesizing sulfonium salts includes providing diaryl iodonium salts in the presence of diaryl sulfides, and exposing the diaryl iodonium salts in the presence of diaryl sulfides to microwave radiation for assisted formation of triaryl sulfonium salt photoacid generators.

The preferred diaryl iodonium salt is selected from at least one of a diphenyl iodonium salt. The preferred diaryl sulfide is selected from at least one of a diphenyl sulfide.

It is also preferred that the diaryl iodonium salt include a fluorene compound that forms the core structure of the triaryl sulfonium salt photoacid generator. The preferred fluorene compound is substituted with at least one of a stilbene motif, a thiophene motif and a nitro group at the 2- and 7-positions; more preferably, the fluorene compound is substituted with a nitro group at the 7-position.

In the preferred method, the reaction time for the synthesis of the sulfonium salt photoacid generator is in a range from approximately 30 seconds to approximately 10 minutes; more preferably, the reaction time for the formation of the sulfonium salt photoacid generator is approximately 6 minutes, which is in a range from approximately 90 to approximately 420 times faster than conventional thermal conditions.

Another preferred method for increasing the efficiency of photoacid generation by inducing intersystem crossing includes a process for synthesizing sulfonium salts including,
selecting a glass reaction vessel suitable for use in a micro-
wave reactor, mixing a fluorene compound that incorporates
a nitro group with a diaryl sulfide in the glass reaction vessel,
purging the mixture with nitrogen gas while stirring, placing
the nitrogen purged mixture in a microwave reactor. setting
the microwave reactor containing the mixture of nitro-group
fluorene compound and diaryl sulfide to closed vessel, stand-
ard mode, maximum pressure 250 psi, maximum tempera-
ture 150 °C., maximum power 60 watts and high speed stirring,
running the reactor for thirty seconds and holding for
30 minutes, removing a nitro-containing, triaryl sulfoxonium salt
that exhibits increased photoacid quantum yield in a range of
from approximately 0.01 to approximately 0.4 times greater
than yields from conventional synthesis.

The preferred method for increasing the efficiency of pho-
toacid generation uses a fluorene compound that incorporates
a nitro group is 9,9-didecyl-2-iodo-7-nitro-9H-fluorene and a
diaryl sulfide that is vinyl sulfide.

Further objects and advantages of this invention will be
apparent from the following detailed description of preferred
embodiments which are illustrated schematically in the
accompanying drawings.

BRIEF DESCRIPTION OF THE FIGURES

FIG. 1 shows a synthesis scheme for a photoacid generator
PAG 7 wherein reactions are carried out both by conventional
methods (Δ) and microwave heating (MW).

FIG. 2 shows a synthesis scheme for a photoacid generator
PAG 11, wherein conventional reactions are carried out at
125 °C. and microwave-assisted synthesis is carried out with
100 watts energy input, closed vessel, standard mode; 40 psi.
FIG. 3 shows a synthesis scheme for a photoacid generator
PAG 13, wherein a first microwave assisted reaction occurs
under reaction conditions including 150 °C.; 100 watts micro-
wave energy; closed vessel, standard mode; 40 psi forming
precursor compound 12 and a second microwave-assisted reaction
occurs in the presence of diphenyl iodonium under reaction conditions including, 125 °C., 150 watts microwave energy,
closed vessel, standard mode; 100 psi forming PAG 13, at 51% yield. Conventional synthesis reactions are carried out at
125 °C.

FIG. 4 shows a synthesis scheme for a photoacid generator
PAG 16, wherein reaction conditions include reacting a pre-
cursor compound 15 in the presence of Cu(II) benzoate and
chlorobenzene at 125 °C. for both conventional and micro-
wave-assisted synthesis, with microwave energy at 100 watts
(W), closed vessel, standard mode; 40 psi. Microwave-assis-
ted synthesis yield is 83% in six minutes; conventional synthesis yield is 71% after 24 hours reaction time.

FIG. 5 shows a synthesis scheme for several photoacid
generators PAG 18, PAG 20, PAG 22 wherein the reaction
conditions include 125 °C. for conventional synthesis and
microwave-assisted synthesis conditions include 100 watts
(W), closed vessel, standard mode; 40 psi; and 125 °C.

FIG. 6 shows a synthesis scheme for sulfoxonium photoacid
generators having a fluorenyl scaffold.

DESCRIPTION OF THE PREFERRED
EMBODIMENTS

Before explaining the disclosed embodiments of the
present invention in detail it is to be understood that the
invention is not limited in its application to the details of the
particular arrangements shown since the invention is capable of other embodiments. Also, the terminology used herein is
for the purpose of description and not of limitation.

It would be useful to discuss the definitions of some words
and abbreviations used herein and their application in
explaining the invention.

"CEM Corporation" is the manufacturer microwave syn-
thesis equipment, including the Discover™ System which is
the smallest instrument on the market today. This synthesizer
is a stand-alone synthesis with a small footprint, and allows
the use of one mL up to 125 mL reaction vessels; the Dis-
cover™ System controls reaction temperatures, pressures,
and stirring speeds.

"R" stands for alkyl radical, such as, but not limited to
dimethyl, diethyl, propyl and the like.

"Standard mode" is a setting on CEM microwave reactors
by means of which the power of microwave radiation that is
delivered to the reaction cavity and is modulated in order to
keep the specified temperature or pressure.

Compounds are numbered herein and are identified in the
listing below:

Compounds number: Chemical name and/or function

1 4-(phenylthio)benzaldehyde, a bis-phenyl sulfide
2 (4-bromomethylphenyl)(phenyl)sulfone, a benzyl bromide
3 an alkylnalene
4 9,9-didecyl-2,7-diformylfluorene, an alkylated diformylfluorene
5 4,4’-(1E,1’E)-2,2’-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-
2,1-diyli)nben(1,4-phenylene)bis(phenyl) sulfinilum, precursor
compound 5
6 diphenyliodonium hexafluorophosphate (V), a reagent.
7 4,4’-(1E,1’E)-2,2’-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-
2,1-diyli)nben(1,4-phenylene)bis(diphenylsulfinilum)
hexafluorophosphate(V), a sulfonium salt and photoacid

generator, PAG 7.
8 phenyl(4-vinylphenyl)sulfane
9 9,9-didecyl-2-iodo-7-nitro-9H-fluorene, a nitro-containing
fluorene compound.
10 (E)-(4,4’-(1E,1’E)-2,2’-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-
2,1-diyli)nben(1,4-phenylene)bis(diphenylsulfinilum)
hexafluorophosphate(V), a sulfonium salt and photoacid

generator, PAG 13.
11 (E)-(4,4’-(1E,1’E)-2,2’-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-
2,1-diyli)nben(1,4-phenylene)bis(diphenylsulfinilum)
hexafluorophosphate(V), a sulfonium salt and photoacid

generator, PAG 11.
12 9,9-didecyl-7-nitro-9H-fluorene-2-yl(phenyl)sulfane
13 (9,9-didecyl-7-nitro-9H-fluorene-2-yl)diphenylsulfonium
hexafluorophosphate(V), PAG 13
14 1-(bromothiophen-2-yl)ethanone, a
brominated acetyliodine derivative
15 (E)-1-(5-bromothiophen-2-yl)3-(4-(phenylthio)phenyl)prop-2-
ene-1-one, precursor 15
16 (E)-(4,4’-(1E,1’E)-2,2’-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-
2,1-diyli)nben(1,4-phenylene)bis(diphenylsulfinilum)
hexafluorophosphate(V), PAG 16
17 An alkylphenyl sulfide
18 A diphenylmethylsulfinilum salt
19 A diaryl sulfide
20 A triaryl sulfinilum salt
21 A diaryl sulfide
22 (4-formylphenyl)diphenylsulfinilum hexafluorophosphate(V), PAG 22

The present invention provides efficient methods for pre-
paring sulfinonium photoacid generators (PAGs) by micro-
wave-assisted reaction of diphenylsulfides in the presence of
diphenyliodonium salts. Several of the PAGs are novel and
have potential for use as 2PA photoinitiators in negative
resists for 3D microfabrication or in optical data storage.
Furthermore, the efficiency of forming these sulfinonium salts
by the microwave-assisted process was evaluated relative to
the conventional thermal reaction of diphenylsulfides in
the presence diphenyliodonium salts. Finally, photoacid quan-
tum yields were determined for several of the novel PAGs.

The new PAGs were designed to exhibit high two-photon
absorbing (2PA) cross sections. Because of its high thermal
and photochemical stability, fluorene was chosen as the core structure of the PAGs, as discussed by Belfield, K. et al. in *Photochem. Photobiol. Sci.* 2004, 3, 138-141. Quite advantageously, fluorene lends itself to ready substitution in its 2-, 7-, and 9-positions. Substituents or thioether motifs were introduced in 2- and 7-positions, in order to extend the π-conjugation. Ultimately, two acceptor groups (triarylsulfonium and nitro) are introduced.

**Example 1**

Microwave-Assisted Synthesis of Sulfonium Salts

As an example of the general procedure for the microwave-assisted preparation of sulfonium salts. The preparation of (4,4′-(1E,1′E)-2,2′-(9,9-didecyl-9H-fluorene-2,7-diyl)bis(ethene-2,1-diyl)bis(4,1-phenylene))bis(diphenylsulfonium) hexafluorophosphate (V), PAC 7.

Materials

Compounds 1 (formyl phenyl phenyl sulfide) and 4 (9,9-didecyl-2,7-diformylfluorene) in FIG. 1 were synthesized according to methods reported in Sivasubramaniam, S. et al. *Indian Journal of Chemistry Section B—Organic Chemistry Including Medicinal Chemistry* 1991, 30, 1148-1149 and Belfield, K. D. et al., in Polymers for Advanced Technologies 2005, 16, 150-155, respectively. Thioureas and phenyl sulfides were used as provided by suppliers. All glassware was flame dried and cooled in a dessicator over calcium chloride. All reactions were carried out under nitrogen atmosphere.

All sulfonium salt synthesis and purification were carried out under yellow light, red light, or in the dark. The sulfonium salt preparation, when carried out by conventional heating methods, was done according literature procedures using chlorobenzene as a solvent.

In a 2 mL glass reaction vessel, purchased from CEM Corporation, microwave reactor manufacturer (heavy walled vessels with corresponding septum designed to withstand up to 3000 psi inside the reaction chamber). 0.21 g (0.24 mmol) of 5, 0.21 g (0.48 mmol) of diphenyliodonium hexafluorophosphate (V), and 0.006 g (5% molar) copper(II) benzoate mixture in 2 mL of chlorobenzene were mixed in the dark in 2 mL of chlorobenzene while being purged with N₂. The reaction was set to closed vessel temperature or pressure model; maximum pressure 40 psi; maximum temperature 150-155 °C. The microwave was set to closed vessel temperature or pressure model; maximum pressure 40 psi; maximum temperature 150-155 °C. The run time (time at which the reaction reaches max temperature or pressure) was set for 20 min. The reaction was monitored by thin layer chromatography (TLC) until most of the starting material had disappeared. Rarely did sulfides convert entirely. In the cases in which the reaction time was extended to try to achieve full conversion, side products seriously complicated the purification hindering the overall yield of the sulfonium salt.

Upon completion, 12 min hold time, the solvent was slowly removed under vacuum. The resulting alcohol was refluxed in HBr (33% in AcOH to obtain the benzyl bromide intermediate, subsequently used to obtain the bisphosphonate. The second branch of the synthesis started from fluorene, alkylation of the 9-position with bromodecane imparted the desired solubility to alkylated fluorene molecule 3, as reported by Belfield, K. et al. *J. Org. Chem.* 2000, 65, 4475-4481.

Bromination with NBS (N-bromosuccinimide) of the 2- and 7-positions of the alkylated fluorene 3 was followed by introduction of the formyl group. A bisformyl fluorene, 9,9-didecyl-2,7-diformylfluorene 4 and the bisphosphonate were coupled via a Wadsworth-Horner-Emmons reaction, yielding exclusively the trans-isomer 5. Bis-sulfide 5 was the precursor for bis-sulfonium salt 7 under both conventional microwave-assisted procedures.

The molecule, PAG 7, exhibited high fluorescence quantum yield (0.80) liming the photocoag quantum yield to (0.03) when excited at 400 nm. The direct photolysis of triarylsulfonium salts has been reported to occur primarily from the first excited singlet state. However, sensitization studies have shown that triplet triarylsulfonium salts are also liable, as reported by Dektar, J. L. et al. *J. Am. Chem. Soc.* 1990, 112, 6004-6015. Consequently, a decision was made to incorporate a nitro group in the structures to induce intersystem crossing.

FIGS. 2 and 3 show the synthesis of sulfonium salts 11 and 13, respectively, using the versatile 2-iodo-7-nitro-9,9-didecylfluorene precursor 9. The versatile 2-iodo-7-nitro-9,9-didecylfluorene precursor 9 was prepared as reported by Belfield et al. in *J. Org. Chem.* 2000, 65, supra. Interestingly, (E)-4-(2-(9,9-didecyl-7-nitro-9H-fluoren-2-yl)vinyl)phenyl(phenyl)sulfane 10 was obtained quantitatively via a microwave-assisted palladium acetate-catalyzed Heck reaction, coupling vinyl sulfide 8 and precursor 9. The yields for both conventional and microwave-assisted methods were the highest for PAG 11 of the entire series. Fluorenylphenylsulfide 12 was prepared from Cu-catalyzed S-arylation of 9 and thiophenol.
In FIG. 2, the synthesis scheme for PAG 11 is carried out by conventional means at a reaction temperature of 125°C. The microwave-assisted synthesis is carried out at 100 watts, closed vessel, standard mode; 40 psi; 125°C. In all of the microwave assisted reactions reported here the maximum established temperature is reached before the maximum established pressure, so the first is the parameter that is used to control the dose of microwave radiation that was delivered to the reaction cavity. Percentages indicate percent yield. The side products of the reactions were not characterized and the percent conversions were not determined because the reaction yields are high enough for these factors to be irrelevant. A step by step detailed description of the reaction conditions of all new compounds can be found in the supplementary information section. Among the conditions stated are reaction temperatures, times, and pressures both for conventional and microwave assisted reactions. In this section you will also find the characterization of each new compound by 1H NMR, 13C NMR and elemental analysis.

In FIG. 3, the synthesis scheme for PAG 13 is carried out as follows:

As a result of incorporating the nitro group in the structures as shown in FIGS. 2 and 3, the fluorescence quantum yield of precursors 10 (in FIG. 2) and 12 (in FIG. 3), and sulfonium salts 11 (in FIG. 2) and 13 (in FIG. 3), were significantly decreased (Table 1), thereby reducing the radiative decay pathway. Nitro-containing sulfonium salt 11 exhibited an increased photoacid quantum yield (Table 1). The photoacid quantum yields were determined by a steady state method in which solutions of the PAGs in acetonitrile were selectively irradiated at the desired wavelength with an excitation source and monochromator of a spectrofluorimeter. Rhodamine B was used as a sensor for photoacid generation, as reported by Pohlers, G. et al. in Chem. Mater., 1997, 9, 1353-1361.

Special care was taken to record the photodecomposition conversion no greater than 5% in order keep secondary photoproduct generation to a minimum. Previously, we determined a number of photochemical quantum yields under one- and two-photon excitation, and found no significant difference relative to excitation mode. Hence, the values of quantum yields in Table 1 are good estimates under 2PA, as reported by Corredor, C. C., et al. in J. Photochem. Photobiol., A, 2006, 184, 177-183.

A diphenyl sulfide, (E)-1-(5-bromoacetylthiophen-2-yl)-3-(4-(phenylthio)phenyl) prop-2-en-1-one 15 was prepared to evaluate how the thiophene and chalcone functionalities would withstand the rapid heating conditions of the microwave-assisted (MW) method. Further demonstrating the versatility of the MW-method. Precursor 15 resulted from a straightforward Claisen condensation of 1-(5-bromoacetylthiophen-2-yl)ethanone, a brominated acetyliophene derivative 14 and formyl phenyl phenyl sultide 1 as shown in FIG. 4. Both functionalities were intact after the microwave assisted reaction was carried out.

In FIG. 4, the synthesis scheme for PAG 16 is carried out as follows: A brominated acetyliophene derivative, 1-(5-bromoacetylthiophen-2-yl)ethanone (14) is reacted with 4-(phenylthio)benzaldehyde (1) in the presence of methanol and potassium hydroxide at room temperature to form a precursor compound, (E)-1-(5-bromoacetylthiophen-2-yl)-3-(4-(phenylthio)phenyl)prop-2-en-1-one (15). The precursor compound (15) is exposed to microwave heating at 125°C. In the presence of diphenyliodonium hexafluorophosphate (V) and methanol the Quantum yield per molecule of PAG 16 is 71% for conventional heating (A) and 83% for microwave heating (MW). Furthermore, we sought to compare the effectiveness of this method when performed on less conjugated sulfides that may be more relevant in UV or deep UV lithographic applications in FIG. 5.

In FIG. 5, a commercially available phenyl sulfide is reacted with diphenyl iodonium hexafluorophosphate (V) to prepare the precursor compound 17. The precursor compound (15) was then reacted with diphenyl iodonium hexafluorophosphate (V) in the presence of a small amount of copper (II) acetate, allowing for microwave heating to a reaction temperature of 125°C. Reagent 17 was chemically similar to the sulfonium salt 13, however the reaction time was extended significantly reduced reactions times for the formation of these sulfonium salts when compared to conventional heating times performed by using reagent 21, as reported in the literature, product 20 and product 22 in Table 2 below.

<table>
<thead>
<tr>
<th>Reagent</th>
<th>Product</th>
<th>MW Time (Yield)</th>
<th>Δ Time (Yield)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>7</td>
<td>6 min (75%)</td>
<td>29 h (42%)</td>
</tr>
<tr>
<td>10</td>
<td>11</td>
<td>8 min (97%)</td>
<td>8 h (87%)</td>
</tr>
<tr>
<td>12</td>
<td>13</td>
<td>10 min (70 h)</td>
<td>70 h (100%)</td>
</tr>
<tr>
<td>15</td>
<td>16</td>
<td>14 min (83%)</td>
<td>24 h (71%)</td>
</tr>
<tr>
<td>17</td>
<td>18</td>
<td>120 s (72%)</td>
<td>3 h (100%)</td>
</tr>
<tr>
<td>21</td>
<td>22</td>
<td>90 s (99%)</td>
<td>3 h (97%)</td>
</tr>
<tr>
<td>8</td>
<td>23</td>
<td>5 s (88%)</td>
<td>3 h (83%)</td>
</tr>
</tbody>
</table>

The reactions that were carried out under conventional conditions, in Table 2, were performed in an oil bath heated to 120° C. at atmospheric pressure; whereas, the microwave-assisted reactions were run at the same temperature in closed vessel mode, reaching pressures no higher than 30 psi. In every case, the reactions that were heated conventionally took significantly longer than the analogous microwave reactions, and microwave-assisted yields were higher.

In conclusion, the present invention provides a new, more favorable methodology for the formation of sulfonium salts via microwave-assisted decomposition of diphenyliodonium salts in the presence of diphenylsulfides. Microwave-assisted reaction times were 90 to 420 times faster, resulting in generally higher yields. In addition, the introduction of groups that favor intersystem crossing is a viable means for increasing the photoacid quantum yield of novel triaryl sulfonium salt photoacid generators (PAGs).

Commercial manufacturers of sulfuric PAGs for use in the photolithography and coatings industries can prepare PAGs faster, easier, and more efficiently, while realizing production energy cost savings. By shortening reaction time from hours, approximately 20 hours at elevated temperatures in a range of 125° C., to minutes, approximately 6-10 minutes, using the process of the present invention, greater efficiencies and lower energy consumption is achieved. Uniformly higher yields than conventional thermal methods are provided by the present invention.

While the invention has been described, disclosed, illustrated and shown in various terms of certain embodiments or modifications which it has presumed in practice, the scope of the invention is not intended to be, nor should it be deemed to be, limited thereby and such other modifications or embodiments as may be suggested by the teachings herein are particularly reserved especially as they fall within the breadth and scope of the claims here appended.

We claim:

1. A method for formation of sulfonium salts comprising the steps of:
   providing a diaryl iodonium salt in the presence of a diaryl sulfide;
   and exposing the diaryl iodonium salt in the presence of the diaryl sulfide to microwave radiation that is delivered to a closed reaction cavity and is modulated for a maximum pressure of approximately 40 psi, a maximum temperature of approximately 125° C. and a maximum power of 100 watts with stirring for assisted formation of a triaryl sulfonium salt photoacid generator wherein the reaction time for the formation of the sulfonium salt photoacid generator is in a range from approximately 30 seconds to approximately 10 minutes.

2. The method of claim 1, wherein the diaryl iodonium salt is selected from at least one of a diphenyl iodonium salt.

3. The method of claim 1, wherein the diaryl sulfide is selected from at least one of a diphenyl sulfide.

4. The method of claim 1, wherein the diaryl iodonium salt further includes a fluorene compound.

5. The method of claim 4, wherein the fluorene compound is substituted with at least one of a stibene motif, a thiophene motif and a nitro group at the 2- and 7-positions.

6. The method of claim 4, wherein the fluorene compound is substituted with a nitro group at the 7-position.

7. The method of claim 1, wherein the reaction time for the formation of the sulfonium salt photoacid generator is in a range from approximately 3 minutes to approximately 10 minutes.

8. The method of claim 7, wherein the reaction time for the formation of the sulfonium salt photoacid generator is approximately 6 minutes.

9. A method for increasing the efficiency of photoacid generation by preparing a nitro-containing sulfonium salt and inducing intersystem crossing by a process consisting essentially of:
   selecting a glass reaction vessel suitable for use in a microwave reactor;
   mixing a fluorene compound that incorporates a nitro group with a diaryl sulfide in the glass reaction vessel to form mixture I;
   purging the mixture I with nitrogen gas while stirring;
   placing the glass reaction vessel containing the nitrogen purged mixture I in the microwave reactor;
   setting the microwave reactor containing the mixture I of nitro-group fluorene compound and diaryl sulfide to closed vessel, standard mode; maximum pressure 250 psi, maximum temperature 150°C, maximum power 60 watts and high speed stirring;
   running the microwave reactor for thirty seconds and holding for 30 minutes;
   removing a nitro-containing, triarylsulfonium salt that exhibits increased photoacid quantum yield in a range of from approximately 0.01 to approximately 0.4 times greater than yields from conventional synthesis.

10. The method of claim 9, wherein the fluorene compound that incorporates a nitro group is 9,9-didecy1-2-iodo-7-nitro-9H-fluorene.

11. The method of claim 9, wherein the diaryl sulfide is vinyl sulfide.

12. The method of claim 4, wherein the fluorene compound included in the diaryl iodonium salt forms the core structure of the triaryl sulfonium salt photoacid generator formed by the microwave radiation assisted reaction of the diaryl iodonium salt including a fluorene compound and the diaryl sulfide.

13. The method of claim 1, wherein the sulfonium salt photoacid generator formed is: PAG 7, which is the compound having the chemical name (4,4'-(1E,1'E)-2,2'-(9,9-didecy1-9H-fluorene-2,7-diy1)bis(ethene-2,1-diyl)bis(4,1-phenylene))bis(diphenylsulfonium) hexafluorophosphate(V).

14. The method of claim 1, wherein the sulfonium salt photoacid generator formed is: PAG 11, which is the compound having the chemical name (E)-(4-(2-(9,9-didecy1-7-nitro-9H-fluorene-2-yl)vinyl)phenyl)diphenylsulfonium hexafluorophosphate(V).

15. The method of claim 1, wherein the sulfonium salt photoacid generator formed is: PAG 13, which is the compound having the chemical name (9,9-didecy1-7-nitro-9H-fluorene-2-yl)diphenylsulfonium hexafluorophosphate(V).

16. The method of claim 1, wherein the sulfonium salt photoacid generator formed is: PAG 16, which is the compound having the chemical name (E)-(4-(3-(4-bromo-naphtho-2-yl)-3-oxoprop-1-enyl)phenyl)diphenylsulfonium hexafluorophosphate(V).

17. The method of claim 1, wherein the sulfonium salt photoacid generator formed is: PAG 22, which is the compound having the chemical name (4-formylphenyl)diphenylsulfonium hexafluorophosphate(V).

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